ALKALOIDS FROM ALSTONIA CONGENSIS

CATHERINE CARON, ALAIN GRAFTIEAUX, GEORGES MASSIOT, LOUISETTE LE MEN-OLIVIER and CLEMENT DELAUDE*

Faculte de Pharmacie (U A au CNRS No. 492), 51 rue Cognacq-Jay, 51096 Reims Cédex, France; *Universite de Liège, Sart-Tilman, 4000 Liège, Belgium

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Abstract—Fifteen alkaloids were found in the root bark, stem bark and leaves of *Alstonia congensis*. They are echitamidine, echitamine, *nor*-echitamine, 17-acetoxy-*nor*-echitamine, akuammicine, 12-methoxyakuammicine, 12-methoxyy-N(4)-methylakuammicine, tubotaiwine, 12-methoxytubotaiwine, angustilobines A and B, 6,7-seco angustilobines A and B, angustilobine B-N(4)-oxide and akuammidine

INTRODUCTION

Monachino's classification of the genus Alstonia retains two African species, A. boonei and A. congensis [1]. These plants are closely related and resemble the widely distributed A. scholaris. Alstonia congensis Engl. has been the object of previous chemical investigations limited to the isolation of echitamine [2, 3], of echitamidine [3, 4] and of triterpenes [5]. A recent report on the isolation of rhazine from the stem bark of A. congensis [6] prompts us to disclose our results on the alkaloid content of the root bark, stem bark and leaves of the plant.

RESULTS AND DISCUSSION

Material was collected from a large ornamental tree growing in the suburbs of Kinshasa, Zaire. Tertiary alkaloids were isolated in the usual fashion [7], with the following yields: 1.55 g/kg (root bark), 0.48 g/kg (stembark) and 2.5 g/kg (leaves) Owing to the small quantities of available material, alkaloids were purified by CC followed by prep. TLC. After the preliminary extraction of the tertiary alkaloids, quaternary ammonium salts were extracted from the remaining solid with *n*-butanol Fifteen pure compounds were obtained from the three parts of the plant, their occurrence and the means of identification are listed in Table 1.

Echitamine (2) is the major alkaloid of the root bark; it is accompanied by two minor bases with the same characteristic Ph-N-C-N- chromophore: nor-echitamine (3) and its 17-O-acetate (4) Compounds 3 and 4 were identified by comparison of their spectra with those of the same compounds available from other Alstonia species [8, 9] Structural elucidation of the antimalarial echitamine has been an area of intense research in the fifties and the problem was solved by X-ray analysis, [10]

To the best of our knowledge, only partial NMR data on 2 are available [11]. To fill this gap we have investigated the ¹H and ¹³C NMR spectra of 2 using 2D techniques; Table 2 gives ¹³C assignments and ¹H NMR is detailed in the Experimental. Quaternarization of N(4) induces ca 10 ppm ¹³C and ca 1 ppm ¹H deshieldings;

N(4)-Me thus appears as a singlet at δ 3.6 instead of 2.24 as reported [11]. Protonated carbons are assigned through a direct C-H correlation and non-protonated carbons through a 'long range' correlation optimized for J=10 Hz This latter experiment allows distinction between C-7 and C-16, which shows 3J coupling with H-14

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Table 1 Occurrence and identification of alkaloids in Alstonia congensis

	Yield (%)								
Alkaloid	Root bark	Stem bark	Leaves	cem	IR	UV	MS	HNMR	¹³ C NMR
Echitamidine 1	3	2		+		+	+		
Echitamine 2	10	5		+	+	+	+	+	+
nor-Echitamine 3	1			+			+	+	
17-O-acetyl-nor-echitamine 4	1		0 25	+		+	+		
Akuammicine 5	1			+		+	+		
12-Methoxyakuammicine 6	0.5	0.5		+		+	+		
12-Methoxy-N(4)-methylakuammicine 7	10				+	+	+	+	+
Tubotaiwine 8	1			+		+	+		
12-Methoxytubotaiwine 9			0.25		+	+	+	+	+
Angustilobine A 10		0.5	3	+		+	+	+	
6,7-Seco-angustilobine A 11			3		+	+	+	+	+
Angustilobine B 12		0.5		+		+	+	+	
Angustilobine-B-N-oxide 13		0.5			+	+	+	+	
6,7-Seco-angustilobine B 14	10		6			+	+	-+-	+
Akuammidine 15		0.3		+		+	+		

Besides echitamidine 1 [12], five alkaloids colour blue upon Ce-IV spraying; among them are the ubiquitous (—)akuammicine (5) and (+)tubotaiwine (8) and their 12-methoxylated derivatives 6 and 9 These latter structures are proposed on the basis of high field ¹H NMR spectra whose lower frequencies parts are superimposable on those of the parent compounds and whose aromatic parts consist of three-spin systems identical with the one ob-

Table 2 ¹³C NMR data for the major alkaloids of *Alstonia* congensis (75 MHz, CDCl₃ except for compounds 2 and 7 whose spectra were measured in CDCl₃-CD₃OD)

C	2	7	9	11	14	
	,					
2	100 0	165 4		132 9		
3	68 6	72 1	45 2	53 6	56 5	
5	620	63 1	53 6			
6	408	40.5	43 2	_		
7	60 7	559	_	101 7	100 7	
8	128 9	133 1		127 7	128	
9	126 3	1141	1121	1200	120 1	
10	1199	123 4	1219	120 5	120 5	
11	128 6	1115	109 9	122 3	122 3	
12	1104	144 5		1107	1108	
13	1466	1318	_	136 2	1377	
14	30 9	280	28 1	24 6	28 9	
15	34 5	28 1	30 6	46 1	45 1	
16	55.4	101 3	-	58 6		
17	64 6	_		69 7	709	
18	14 1	132	115	1153	69 2	
19	1299	128 7	23 7	141 1	123 5	
20	1315	128 7	40 8	84 2	1368	
21	65 4	64 3	65.5	61.3	66 8	
N-Me	49 1	49 9		46 8	46 4	
C=O	1726	166 6	_			
OMe	512	51 5	51 2	529	528	
ArOMe	559	55 6				

served in the spectrum of 12-methoxy compactinervine [7] Analysis of the ¹³C NMR spectrum of 9 and comparison with literature data [13] confirm the hypothesis. Compound 6 has previously been isolated from *Vinca ervincea* [14] and prepared by partial synthesis [7] To the best of our knowledge 12-methoxytubotaiwine is described here for the first time and is only the second derivative of tubotaiwine isolated until now [15]

Compound 7 is a quaternary alkaloid obtained from butanol extracts Its UV spectrum displays three maxima at 225, 289 and 333 nm and is reminiscent of the UV spectra of 6 and of 9 The ¹H NMR spectrum of 7 is fully assignable by means of a COSY experiment, it shows signals for an ethylidene chain with allylic and homoallylic couplings to a methylene, signals for a tryptamine CH₂-CH₂ unit and signals for a CH(3)-CH₂(14) -CH(15) unit Three methyls are also characterized by singlets at δ 3 87, 3 8 and 3 76 (¹H NMR) and by signals at 556, 515 and 499 ppm (13C NMR), they are thus assigned to a methyl ether, a methyl ester and a quaternary N-methyl group, respectively. The overall aspect of the ¹H NMR spectrum of 7 suggests a parenthood with 12-methoxyakuammicine Differences regard protons adjacent to nitrogen and may be explained by quaternarization of N(4) by a methyl group. This is also demonstrated by the ¹³C NMR spectrum (Table 2). Beside the three aromatic indole protons, four coupling resonances appear at low field in the 1H NMR spectrum, they gradually disappear during repeated TLC purification using Verpoorte's ammonium nitrate solvent mixtures [16] These signals belong to the negatively charged counterion and are tentatively assigned to a nicotinate ion according to the ¹³C NMR spectrum. The ¹H NMR spectra of native 7, of its chloride and nitrate forms show pronounced chemical shift differences, thus implying that in CDCl₃ and in CDCl₃-CD₃OD mixtures, the molecules form a tight ion pair.

A series of vallesamine derivatives is also present in all parts of the plant, they are angustilobines A and B (10 and 12) and 6,7-secoangustilobine B 14 isolated from two asian Alstonia species, A pneumatophora and A angusti-

loba [9, 17]. Angustilobine B has been independently isolated from A. scholaris and named alstonamine [18]. Alkaloid 11 is a novel vallesamine derivative, isomeric with 14 according to its mass spectrum and for which we propose a structure of 6,7-secoangustilobine A. The ¹H NMR spectrum of 11 shows four aromatic protons plus a singlet at δ 6.39, long-range coupling to the indole NH and featuring H-7 This spectrum also shows an isolated three-spin system for a vinyl group, two AX pairs of doublets (CH₂-17 and CH₂-21) and a five-spin system assigned to the CH₂-3 CH₂-14 CH-15 unit. Observation of signals for a methyl ester ($\delta 3.6$) and for a N-methyl group (δ 2.24) leads to structure 11 with stereocenters configurations identical to those proposed for angustilobine A. In compounds 11 and 14, opening of the eightmembered ring brings strain release and deshieldings of the ¹³C resonances of C-3 and C-21 (suppression of y effects). These vallesamines are accompanied by small quantities of the N(4)-oxide of angustilobine B whose structure is secured by a synthesis from the parent amine. The last compound identified in this study is akuammidine (15) [19] which sometimes is named rhazine [6].

Comparison of the alkaloid contents of A. congensis and of A. boonei does not allow answering the question of their being identical or not. Both species yield echitamine and echitamidine as major alkaloids as well as akuammidine [20]. A boonei also contains voacangine, the sole type III indole alkaloid isolated from an Alstonia species [20] and derivatives of N(1)-formyl echitamidine [21, 22] which were not found in the present study.

EXPERIMENTAL

Extraction and separation Dried powdered root bark (245 g) was wetted with 200 ml of NH₄OH half dild in H₂O and lixiviated overnight with 2.51 of EtOAc. The organic soln was extracted with 2% H₂SO₄ and the aq. phase made alkaline with NH₄OH and extracted with CHCl₃ The CHCl₃ layers were dried (Na₂SO₄) and evapd in vacuo to give 0.381 g of crude alkaloid mixt. (AM). The alkaline aq phase was acidified to pH 6 with HOAc and extracted with n-BuOH, drying and evapn of solvent yielded 0.22 g of extract

Crude AM was purified by CC on 12 g of silica gel packed in CHCl₃; 10 ml fractions were collected Elution was performed with CHCl₃ 400 ml, CHCl₃-MeOH (99.1) 200 ml, (49.1) 280 ml and (19.1) 300 ml

Alkaloids 1, 5 and 6 were in frs 63-70, alkaloids 8 in frs 71-90, 3 and 4 in frs 91-115 and 14 in frs 125-140 Alkaloids 2 and 7 were sepd by prep TLC (MeOH-02 M NH₄NO₃) Alkaloids from the leaves and stem bark were isolated and sepd in a similar fashion Polarity order of the alkaloids is angustilobine A, 6,7-secoangustilobine A, 17-acetoxy-nor-echitamine, 12-methoxy-tubotaiwine, angustilobine B and 6,7-seco-angustilobine B

Echitamine 2 ¹H NMR (300 MHz CDCl₃, CH₃OD) 7 6 (d, J = 7 Hz, H-9), 6.95 (t, J = 7 Hz, H-11), 6.65 (m, H-10+H-12), 5.62 (br q, J = 7 Hz, H-19), 4.45 (br d, J = 14 Hz, H-21), 4.35 (dd, J = 11, 6 Hz, H-3), 3.87 (br d, H-15), 3.68 (s, OMe), 3.24 (s, N-Me), 2.55 (ddd, J = 7, 11, 16 Hz, H-14), 2.32 (dt, J = 7, 10 Hz, H-6), 2.05 (dd, J = 7, 10 Hz, H-6), 1.69 (dd, J = 1.5, 7 Hz, Me-18), 1.46 (dd, J = 4, 11 Hz, H-14)

12-Methoxy-N(4)-methylakuammicine 7 (CR blue), $[\alpha]_D = +70^{\circ}$ (CHCl₃, c 0.15), UV $\lambda_{\rm max}^{\rm MeOH}$ nm 225, 282, 289, 333; $\rm IR\,\nu_{\rm max}^{\rm CHCl_3}\,\rm cm^{-1}$ 3350, 1700, 1660, 1580, 1400, MS m/z (rel. int) 367 [M]⁺, 5), 366 (10), 352 (15), 322 (10), 294 (25), 277 (15), 264 (20), 188 (20), 122 (100), 121 (70), $^{\rm 1}H$ NMR (300 MHz, CDCl₃)

8,8 (br s), 8.68 (br s), 7.95 (d, J=7 Hz), 7 55 (d, J=7 Hz), 7 2 (t, J=7 Hz), 6 95 (m, 2H), 6 8 (m, 2H), 5.8 (q, J=7 Hz, H-13), 5.05 (br s, H-3), 4 5 (m, H-5), 4.32 (d, J=13 Hz, H-21), 4.12 (m, H-5+H-15), 3 93 (d, J=13 Hz, H-21), 3.87 (s, Ar OMe), 3 8 (s, CO₂Me), 3 76 (s, N-Me), 2 9 (dt, J=7, 13 Hz, H-6), 2.45 (br d, J=12 Hz, H-14), 2.25 (dd, J=7, 13 Hz, H-6), 1 8 (d, J=7 Hz, Me-18), 1.6 (br d, J=12 Hz, H-14)

12-Methoxytubotaiwine 9. (CR blue); $[\alpha]_D = +305$ (CHCl₃, c 0 15); UV λ_{max}^{MeOH} nm· 210, 291, 333; IR $\nu_{max}^{CHCl_3}$ cm⁻¹ 3370, 2930, 1730, 1675, 1610, 1490, 1460, 1430, 1265, 1230, 1160, 1100, 1030; MS m/z (rel int) 355 (90), 354 ([M] +, 100), 339 (10), 324 (20), 297 (30), 259 (80), 135 (60), 124 (100), 122 (90), 107 (30), 98 (35), 95 (40), 71 (95), 1 H NMR (300 MHz, CDCl₃)· 8.75 (s, NH), 6 85 (m, 2H), 6 72 (m, 1H), 3.85 (s, Ar OMe), 3.78 (s, CO₂Me), 3 15 (s, 2H), 2.85 (s, 2H), 2 52 (s, 2H, 2 7, 11 Hz, H-3), 2 0 (s, 2H-20), 1.8 (s, 2H), 0.85 (s, 2H-19), 0 7 (s, 3 7 Hz, Me-18).

6,7-Secoangustilobine A 11. (CR grey turning to pink after 2 days), $[\alpha]_D = +78^\circ$ (CHCl₃, c 0 3)· UV λ_{max}^{MeOH} nm 225, 274, 282, 290, IR $\nu_{max}^{CHCl_3}$ cm $^{-1}$ · 3350, 1730, 1450, 1240, 760; MS m/z (rel. int.): 340 ([M] $^+$, 60), 281 (20), 201 (70), 154 (20), 122 (100); 1 H NMR (300 MHz, CDCl₃), 8 4 (s, N–H), 7 55 (d, J=8 Hz, H-9), 7.3 (d, J=8 Hz, H-12), 7.15 (t, J=8 Hz, H-11), 7 08 (t, J=8 Hz, H-10), 6 4 (br s, H-7), 5 65 (dd, J=10, 17 Hz, H-19), 5 45 (dd, J=17, 1.8 Hz, H-18), 5 15 (dd, J=10, 1 8 Hz, H-18), 4 95 (d, J=9 Hz, H-17), 4 62 (d, J=9 Hz, H-17), 3.6 (s, CO₂Me), 2 9 (dd, J=12, 6 Hz, H-15), 2.8 (br d, J=13 Hz, H-21), 2.7 (br d, J=11 Hz, H-3), 2.23 (s, N–Me), 2 01 (d, J=13 Hz, H-21), 1.85 (dt, J=1.8, 10 1 Hz, H-3), 1 4 (m, H-14), 1 2 (m, H-14).

Angustilobine B-N-oxide 13. (CR grey); $[\alpha]_D = +97^\circ$ (CHCl₃, c 0 2); UV λ_{max}^{MOH} nm: 223, 285, 292; IR $\nu_{max}^{CHCl_3}$ cm⁻¹. 3400, 1730; MS m/z (rel int) 353 ([M]⁺, 30), 338 (10), 307 (20), 294 (40), 279 (25), 265 (40), 263 (30), 251 (30), 122 (100), ¹H NMR (300 MHz, CDCl₃)· 7 5–7 1 (m, 4H), 5 5 (br s, H-19), 4.85 (d, J = 17 Hz, H-6), 4 53 (br d, J = 16 Hz, H-18), 4.56 (d, J = 13 Hz, H-17), 4.23 (dd, J = 16, 4 Hz, H-18), 4.0 (d, J = 17 Hz, H-6), 3.87 (s, CO₂Me), 2.05 (m, H-14), 1 72 (m, H-14)

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REFERENCES

- 1 Monachino, J (1949) Pacific Sc III, 133
- 2 Goodson, J. A. (1932) J Chem. Soc. 2627.
- 3 Monseur, X. and Van Bever, M. L (1955) J. Pharm. Belg. 10, 93
- 4 Prista, L. N., Ferreira, M. A., Alves, A. C. and Roque, A. S. (1965) Garcia Orta 13, 571
- 5 Ferreira, M. A., Alves, A. C., Prista, L. N. and Cruz, M. A. (1968) Garcia Orta 16, 31.
- 6. Banerji, A and Jana, S (1986) J Indian Chem. Soc. 63, 449.
- 7 Legseir, B, Cherif, A, Richard, B, Pusset, J., Labarre, S., Massiot, G and Le Men-Olivier, L. (1986) Phytochemistry 25, 1735
- 8 Boonchuay, W. and Court, W. E (1976) Phytochemistry 15, 821
- 9 Ravao, T (1985). Ph.D thesis, University of Reims.
- 10 Hamilton, J A, Hamor, T. A., Robertson, J. M and Sim, G A (1961) Proc. Chem Soc 63
- 11 Conroy, H, Bernasconi, R., Brook, P R, Ikan, R., Kurtz, R. and Robinson, K W (1960) Tetrahedron Letters 1.
- 12 Zeches, M, Ravao, T, Richard, B, Massiot, G., Le Men-

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Olivier, L, Guilhem, J and Pascard, C (1984) Tetrahedron Letters 25, 659

- 13 Verpoorte, R., Van Beck, T. A., Riegman, R. L. M., Hyland, P. I. and Bisset, N. G. (1984) Org. Magn. Reson. 22, 328.
- 14 Yagudaev, M. R. (1983) Khun Prir Soedin 210.
- 15 Lounasmaa, M and Somersalo, P (1986) Fortschr Chem. Org Naturst 50, 27
- 16 Verpoorte, R. and Baerbeim Svendsen, (1976) J. Chromat. 124, 152
- 17 Zèches, M., Ravao, T., Richard, R., Massiot, G., Le Men-Olivier, L. and Verpoorte, R. (1987) J. Nat. Prod. 50, 714
- 18 Atta-Ur-Rahman and Alvi, K. A. (1987) Phytochemistry 26, 2139
- 19 Janot, M.-M., Le Men, I., Gosset, I. and Lévy, I. (1962) Bull. Soc. Chim. Fr. 1079
- Croquelois, G., Kunesch, N., Debray, M. and Poisson, I. (1972) Plant Med. Phyt. 6, 122.
- 21 Oguakwa, J. U., Galeffi, C., Messana, I., Patamia, M., Nicolleti, M. and Marini-Bettolo, G. B. (1983). Gazz. Chim. It. 113, 533
- 22 Oguakwa, I. U. (1984) Phytochemistry 23, 2708.